# INSECT ANTIFEEDANT ELEMANOLIDE LACTONES FROM VERNONIA AMYGDALINA

## IRAJ GANJIAN,\* ISAO KUBO† and PAWEL FLUDZINSKI‡

Division of Entomology and Parasitology, College of Natural Resources, University of California, Berkeley, CA 94720, U.S.A.; 
‡Department of Chemistry, University of Rochester, Rochester, NY 14627, U.S.A.

(Revised received 7 March 1983)

Key Word Index-Vernonia amygdalina; Compositae; insect antifeedant; elemanolide; 11,13-dihydrovernodalin.

Abstract—Chemical investigation of insect antifeedants from the bitter tasting leaves of *Vernonia amygdalina* by the application of semi-preparative reversed phase HPLC has led to the isolation and characterization of vernodalin, vernodalol and a new sesquiterpene lactone, 11,13-dihydrovernodalin. This new compound exhibited *in vitro* cytotoxicity and antifeedant activity against the African armyworm *Spodoptera exempta*. The structure of this new antifeedant was determined by spectroscopic data, and the assignment of the methyl group at the  $\alpha$ -position on C-11 was based on the <sup>1</sup>H NMR long range coupling constants, as well as successive large J values.

#### INTRODUCTION

In our continuing search for naturally occurring insect antifeedants we have examined the bitter tasting leaves of Vernonia amygdalina (Compositae) which are used as a folk medicine to cure fever in East Africa [1]. The two elemanolide lactones, namely vernodalin (1) [2] and vernodalol (2) [3], were previously isolated from this tropical plant as cytotoxic principles. A number of elemanolide lactones have been isolated from Vernonia species [4]. Due to the interest in antitumor activity, much attention has been focused on the synthesis of vernolepintype sesquiterpenoids [5–8].

This paper describes the isolation of insect antifeedant sesquiterpene lactones, mainly by semi-preparative HPLC, which resulted in the separation of vernodalin, vernodalol and a new antifeedant elemanolide lactone, 11,13-dihydrovernodalin.

#### RESULTS AND DISCUSSION

In addition to the two aforementioned known sesquiterpene lactones 1 and 2, a new elemanolide lactone, 11,13-dihydrovernodalin (3), has been isolated from the ether extract of the dried leaves of the East African medicinal plant V. amygdalina as an antifeedant against the African armyworm S. exempta with a leaf-disk assay [9]. Interestingly, despite the lack of an exocyclic conjugated double bond on the  $\gamma$ -lactone moiety, this new sesquiterpene lactone 3 exhibited in vitro cytotoxicity which is almost comparable to vernodalin [2]. The isolation of the antifeedant sesquiterpene lactones was done through monitoring with the leaf disk assay.

After a preliminary separation of the crude extract by column chromatography (silica gel), the corresponding active fraction was submitted to HPLC for further purification on a reversed-phase semi-preparative column. The order of elution under the applied HPLC conditions was: vernodalin 20.6 min; 11,13-dihydrovernodalin 22.8 min; and vernodalol 26.5 min with a peak area ratio of 1:4:1, respectively. Fractions were collected and identification of each pure compound was based on spectroscopic data.

11,13-Dihydrovernodalin, colorless oil, gave the following spectroscopic results; CIMS 363 [M+1]<sup>+</sup> and EIMS showed prominent peaks at m/z (rel. int.) 278 (2.6), 260 (6.5), 248 (38), 230 (100) and 85 (67). Except for the ion at m/z 85 [-CO-C(CH<sub>2</sub>OH)=CH<sub>2</sub>]<sup>+</sup>, each of the other fragment ions was 2 amu higher than the corresponding peak for vernodalin. The UV spectrum showed only end absorption. The IR spectrum displayed characteristic bands at 3400 (hydroxyl group), 1780 ( $\gamma$ -lactone moiety), 1730 (shoulder,  $\delta$ -lactone moiety), 1720 (carboxyl group of hydroxymethacrylate side chain), 1690 (shoulder, conjugated double bond) and 1635 cm<sup>-1</sup> (vinyl group).

The <sup>1</sup>H NMR signals are summarized in Table 1. Assignment of the methyl group at the  $\alpha$ -position of C-11 of the  $\gamma$ -lactone moiety, rather than at C-4 of the  $\delta$ -lactone ring, was based on the coupling constant patterns for 5-H ( $\delta$ 2.90,  $J_{5,14} = 1.8$  Hz, W-type coupling), 7-H ( $\delta$ 2.04) and 11-H ( $\delta$ 2.64) with large J values. This was also supported

<sup>\*</sup>Present address: Department of Chemistry, Herbert H. Lehman College of the City University of New York, Bronx, NY 10468, U.S.A.

<sup>†</sup>To whom correspondence should be addressed.

2526 I. Ganjian et al.

Table 1. <sup>1</sup>H NMR spectral data of 11,13-dihydrovernodalin in CDCl<sub>3</sub>

Assignment	Chemical shift	( $\delta$ ) Coupling constant (Hz)
H-1	1H 5.73	dd, J = 10.8, 17.7
H-2	1H 5.31	dd, J = 2, 17.7
	1H 5.29	dd, J = 2, 10.8
H-5	1H 2.90	dddd, $J = 0.8, 1, 1.8, 11.0$
H-6	1H 4.07	dd, J = 11, 11
H-7	1H 2.04	ddd, $J = 11, 11, 12.1$
H-8	1H 5.12	ddd, $J = 4.6$ , $10.8$ , $11$
H-9	1Hα 1.71	dd, $J = 10.8$ , 14.2
	1Hβ 2.1	dd, $J = 4.6$ , 14.2
H-11	1H 2.64	dq, $J = 6.9$ , 12.1
H-13	3H 1.42	d, J = 6.9
H-14	1Hα 4.28	dd, $J = 1.8$ , 12.1
	1Hβ 4.52	d, J = 12.1
H-15	1H 6.75	dd, J = 0.8, 1
	1H 5.94	dd, $J = 1, 1$
H-3'	2H 4.35	br s
H-4'	1H 6.27	dd, J = 0.8, 1
	1H 5.96	dd, $J = 0.8$ , 1

Table 2. <sup>13</sup>C NMR spectral data of 11,13-dihydrovernodalin in CDCl<sub>3</sub>

Carbon	δ	Carbon	δ
C-1	139.2	C-11	45.7
C-2	116.1	C-12	176.8
C-3	164.5	C-13	13.8
C-4	129.8	C-14	70.0
C-5	40.5	C-15	134.8
C-6	77.0	C-1'	162.9
C-7	53.8	C-2'	138.4
C-8	68.9	C-3′	60.8
C-9	38.0	C-4'	125.8
C-10	40.0		

by the 5-H (dddd, J = 0.8, 1.0, 1.8, 12 Hz) and 7-H (ddd, J = 11, 11, 12 Hz) multiplicities. The <sup>13</sup>C NMR data are tabulated in Table 2. The signals due to C-5 and C-10 overlapped to give a relatively intense peak. The assignment of each peak was based on the observed multiplicity in a single-frequency off-resonance decoupled spectrum.

### **EXPERIMENTAL**

<sup>1</sup>H NMR spectra were recorded on a Bruker FT-400 instru-

ment using TMS as an internal standard in CDCl<sub>3</sub>. <sup>13</sup>C NMR spectra were determined on JEOLCO PH-100 instrument. MS were obtained with a Hitachi Perkin–Elmer, model RMU-6D, single-focusing spectrometer. UV spectra were recorded on a Hitachi UV-100-80 spectrophotometer with 10 mm cell. IR spectra were taken on a Jasco IRA-1 Grating Infrared Spectrophotometer.

Sample preparation. The air-dried leaves (450 g) of V. amygdalina, collected near Nairobi, Kenya, were ground and extracted with Et2O and the extracts were concentrated under red. pres. to a thick oil. The crude material (4 g) was chromatographed on 500 g silica gel. After eluting with 5 l. of CHCl<sub>3</sub> further elution with 1.51. CHCl3-MeOH (99:1) resulted in the separation of an antifeedant active fraction. This fraction (0.5 g) was dissolved in MeOH and after filtration subjected to further purification by reversed-phase semi-preparative HPLC. Mobile phase, MeOH- $\rm H_2O~(1:1)~at~3~ml/min;$  Whatman  $C_{18}$  semi-preparative stainlesssteel column (50 cm × 9.4 mm i.d.) equipped with Whatman stainless-steel guard column (7 cm × 2.1 mm i.d.) packed with pellicular Co: Pell ODS; typical column head pressures under these conditions were 157 bar. Solvents were thoroughly degassed by applying a helium degassing system. The compounds were monitored by UV absorption at a wavelength of 220 nm.

Acknowledgements—We thank Professor Koji Nakanishi and Professor Andrew S. Kende for helpful discussions and Mr. Andrew Chapya for plant collection.

#### REFERENCES

- Kokwaro, J. O. (1976) Medicinal Plants of East Africa, p. 72. East African Literature Bureau, Nairobi.
- Kupchan, S. M., Hemingway, R. J., Karim, A. and Werner, D. (1969) J. Org. Chem. 34, 3908.
- Asaka, Y., Kubota, T. and Kulkarni, A. B. (1977) Phytochemistry 16, 1838.
- Mabry, T. J., Gill, J. E., Burnett, W. C. and Jones, S. B., Jr. in (1977) Host Plant Resistance to Pests (Hedin, P. A., ed.), A.C.S. Symposium Series No. 62, pp. 179–184. American Chemical Society, Washington, D.C.
- Grieco, P. A., Nishizawa, M., Oguri, T., Burke, S. D. and Marionvic, N. (1977) J. Am. Chem. Soc. 99, 5773.
- Danishefsky, S., Schuda, P. F., Kitahara, T. and Etheredge, S. J. (1977) J. Am. Chem. Soc. 99, 6066.
- Iio, H., Isobe, M., Kawai, T. and Goto, T. (1978) J. Am. Chem. Soc. 100, 1940
- 8. Iio, H., Isobe, M., Kawai, T. and Goto, T. (1979) Tetrahedron
- Kubo, I. and Nakanishi, K. (1977) in Host Plant Resistance to Pests (Hedin, P. A., ed.), A.C.S. Symposium Series No. 62, pp. 165–178. American Chemical Society, Washington, D.C.